

Chemical Characteristics of Air Pollution Outflow from North America: Early Insights from Nova Scotia 2004



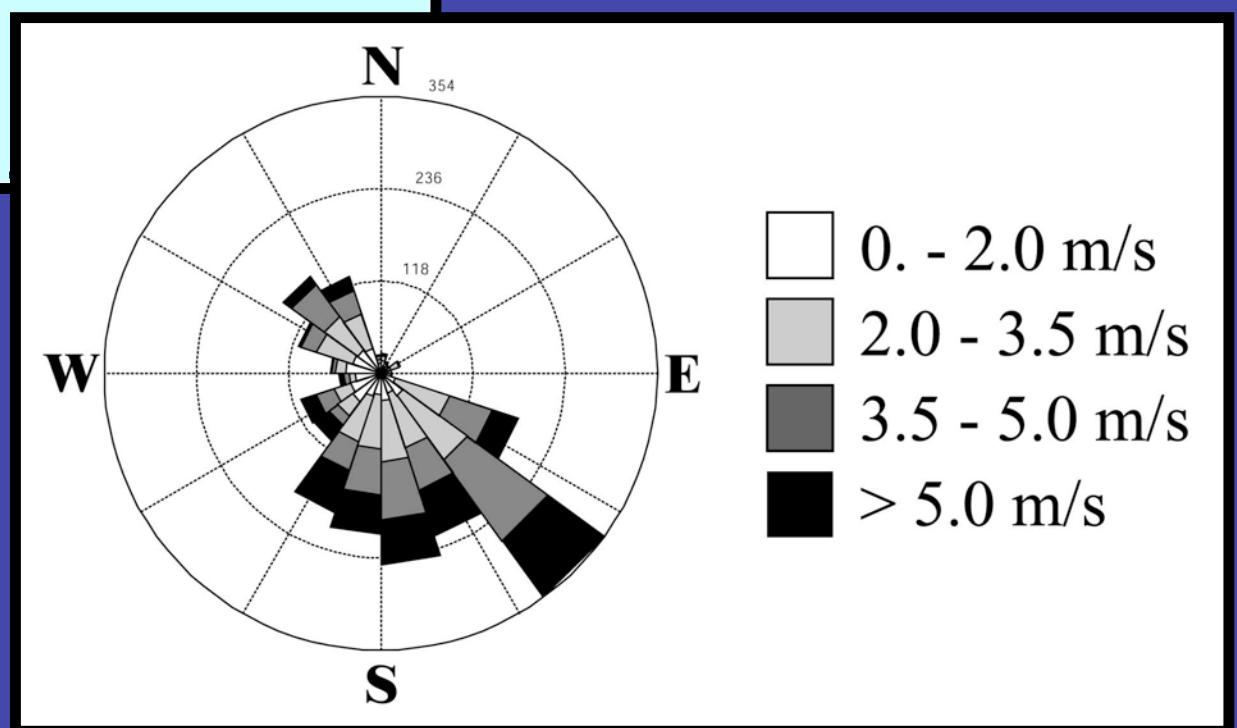
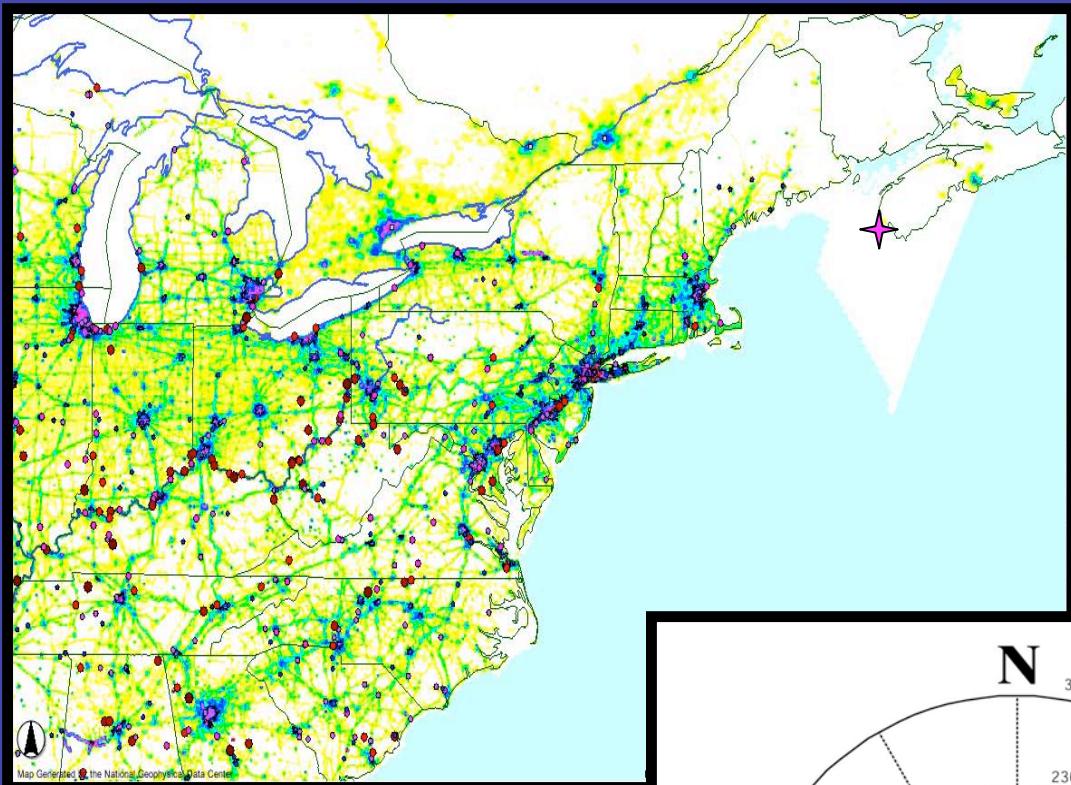
Allen Goldstein

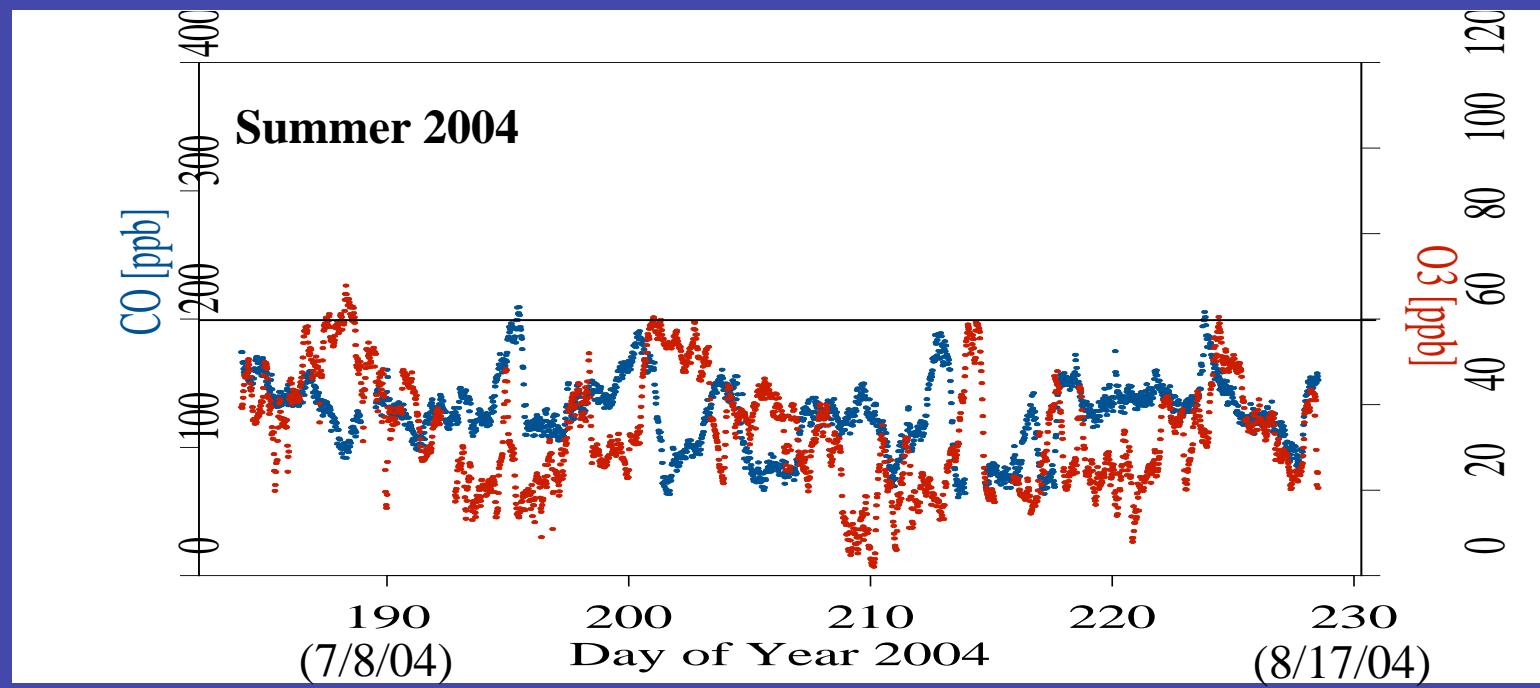
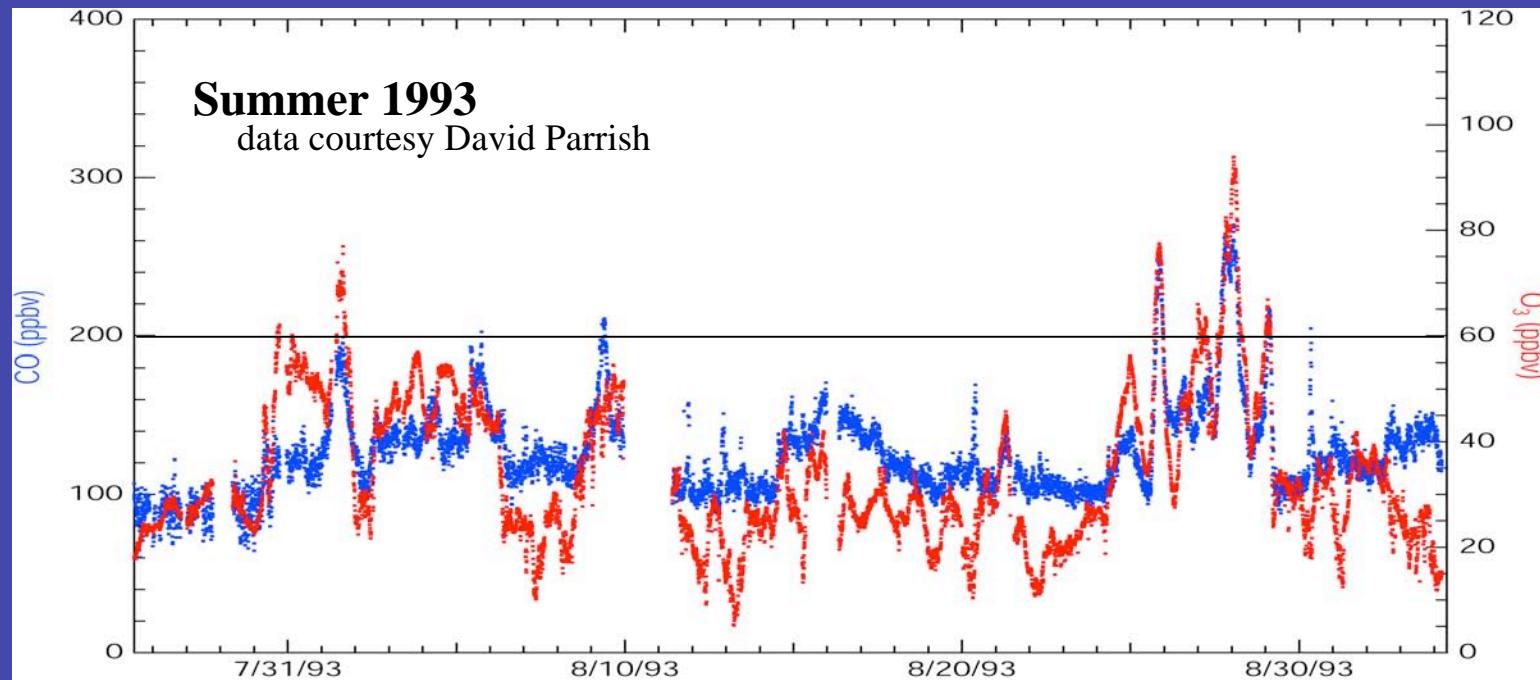
University of California, Berkeley

(MOSTLY PREPARED BY DYLAN MILLET)

Outline

1. Site description & comparison to 1993
2. What were the dominant processes/sources impacting atmospheric composition at CP?
3. What can the combination of VOC/TAG/AMS data tell us about organic aerosol sources & chemistry?
4. What can we say about the composition & chemistry of pollution outflow?





NARE 1993

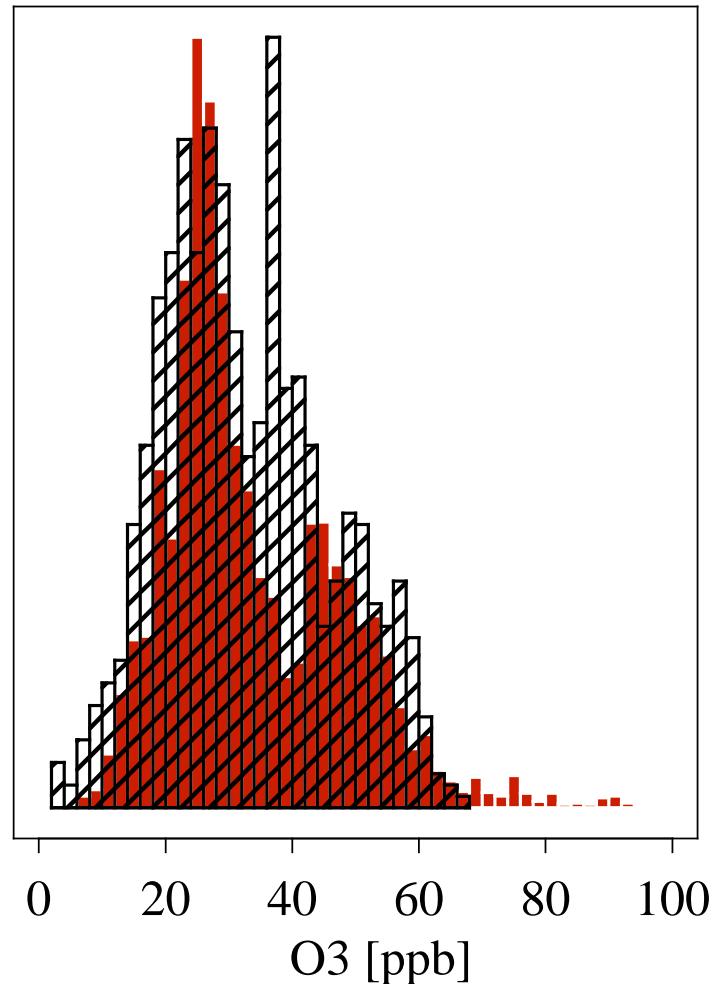
MEAN O₃ (PPB): 34 +/- 14 (RANGE: 5-94)

MEDIAN O₃ (PPB): 30

ICARTT 2004

MEAN O₃ (PPB): 33 +/- 13 (RANGE: 2-68)

MEDIAN O₃ (PPB): 31



NARE 1993

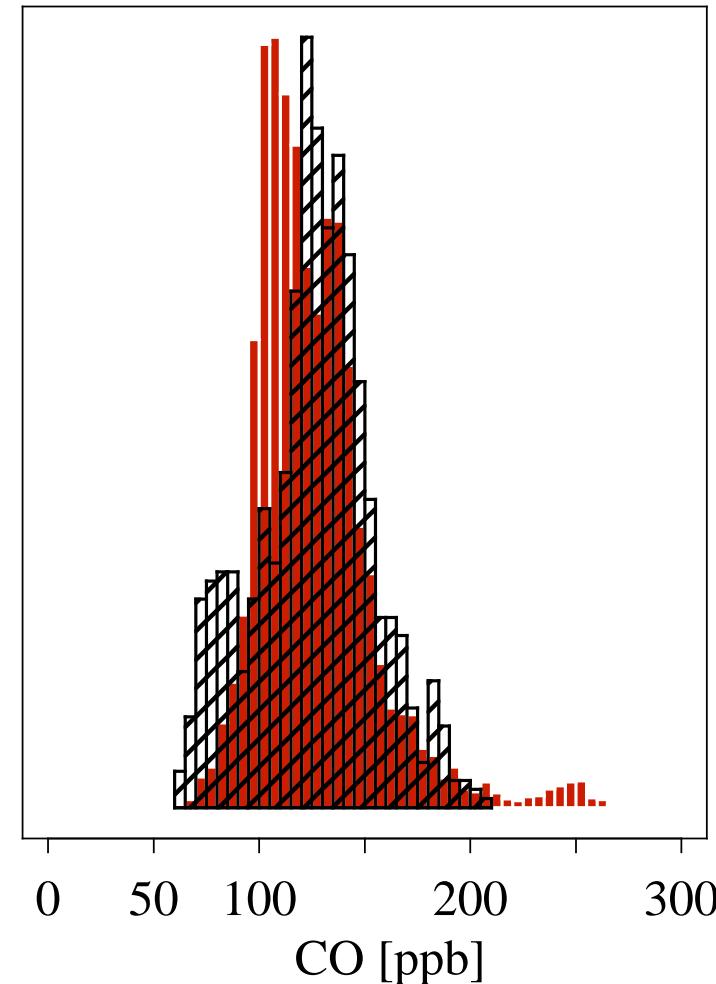
MEAN CO (PPB): 126 +/- 28 (RANGE: 60-270)

MEDIAN CO (PPB): 121

ICARTT 2004

MEAN CO (PPB): 126 +/- 28 (RANGE: 63-206)

MEDIAN CO (PPB): 128



Factor Analysis

F1: US dominated

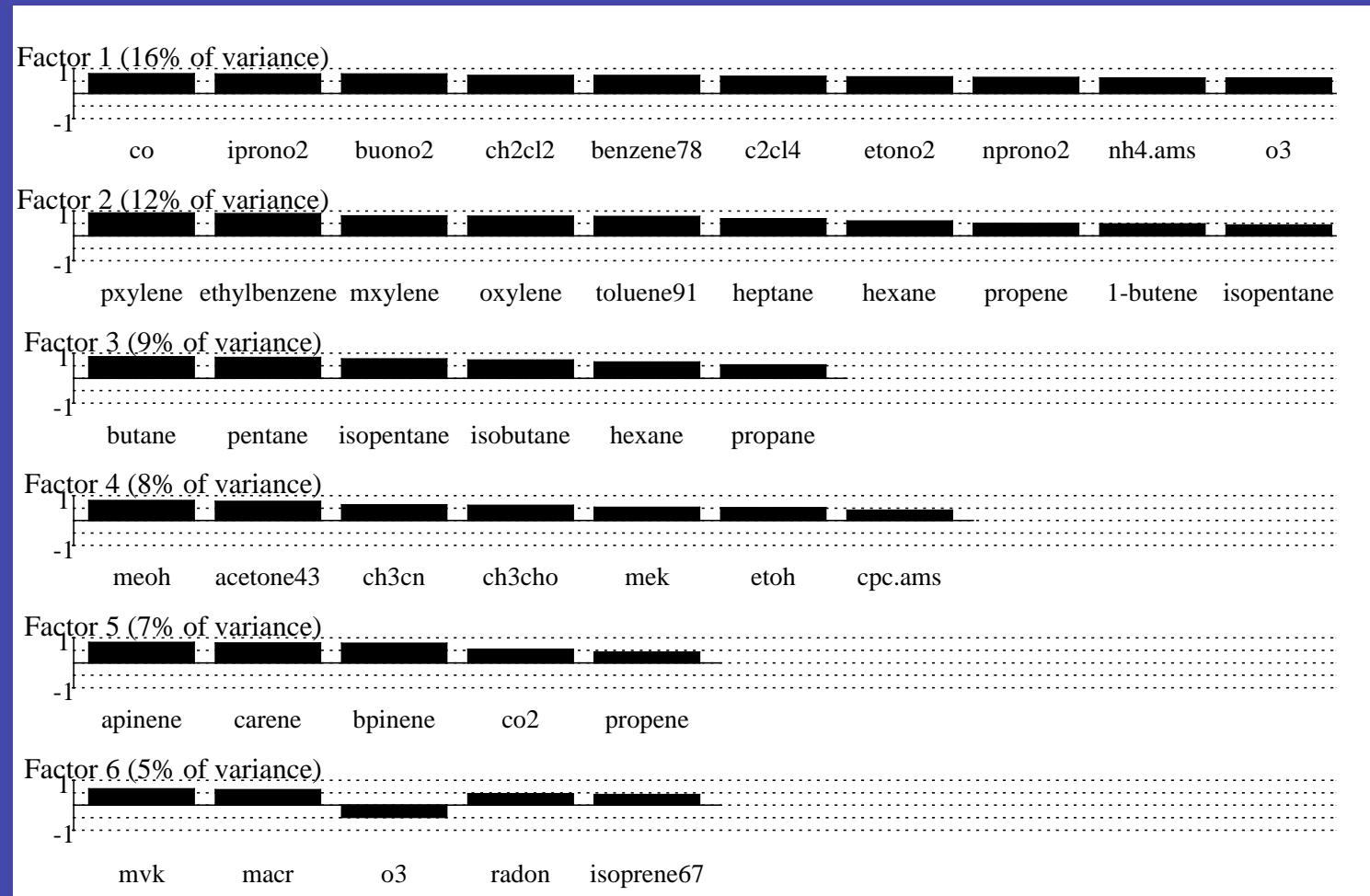
F2: Local combustion

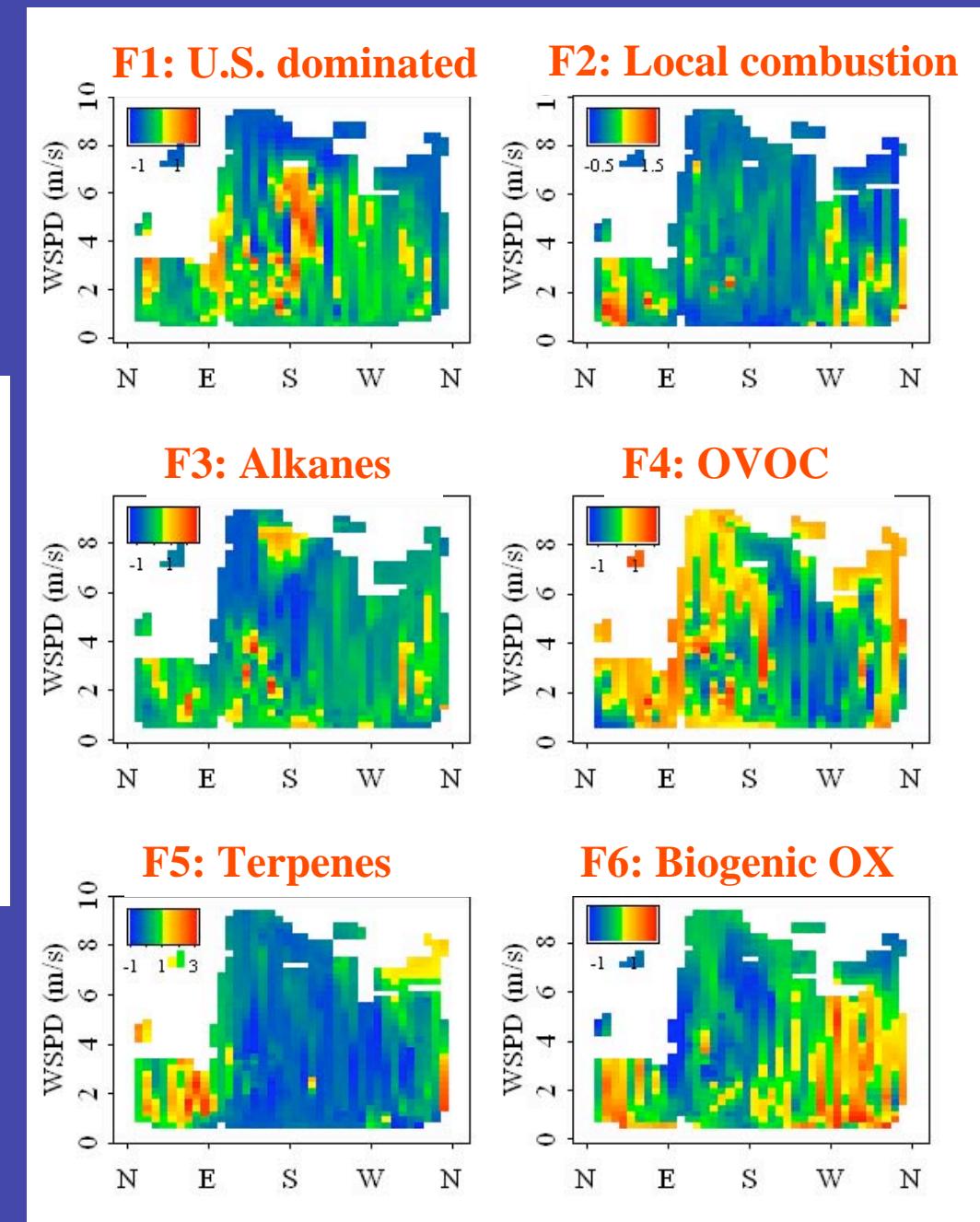
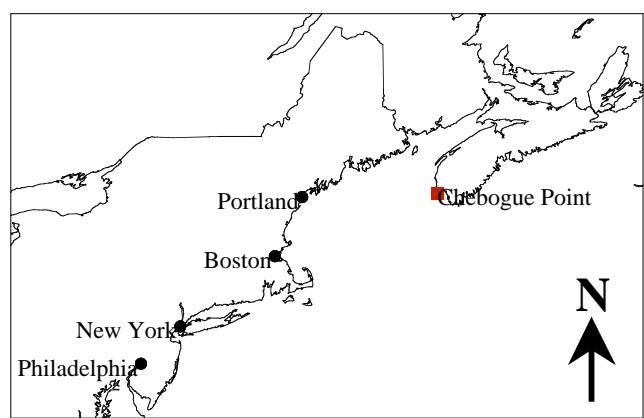
F3: Alkanes

F4: OVOC

F5: Terpenes

F6: Biogenic OX

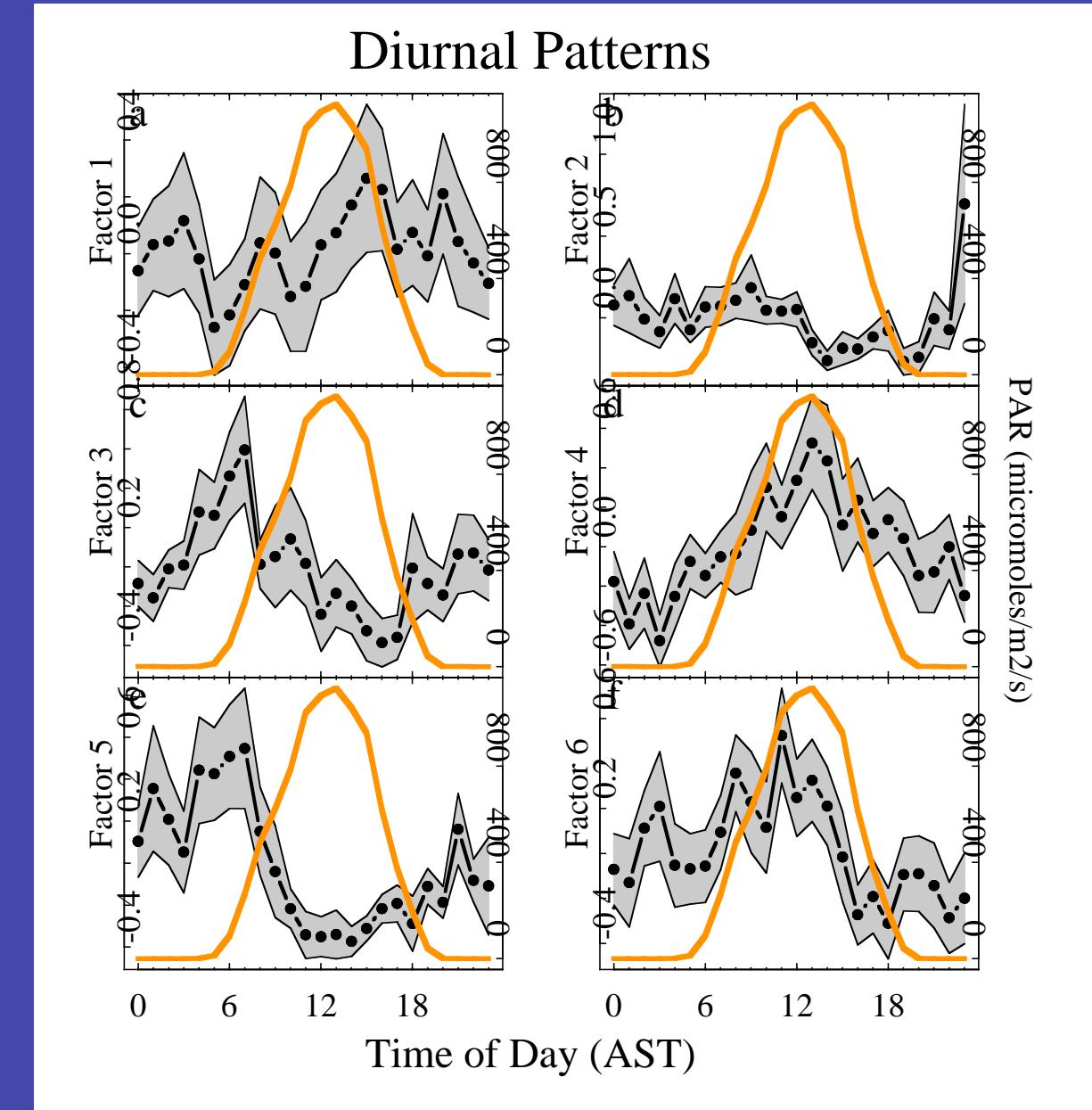




F1: U.S.

F3: Alkanes

F5: Terpenes



F2: Local combustion

F4: OVOC

F6: Biogenic OX

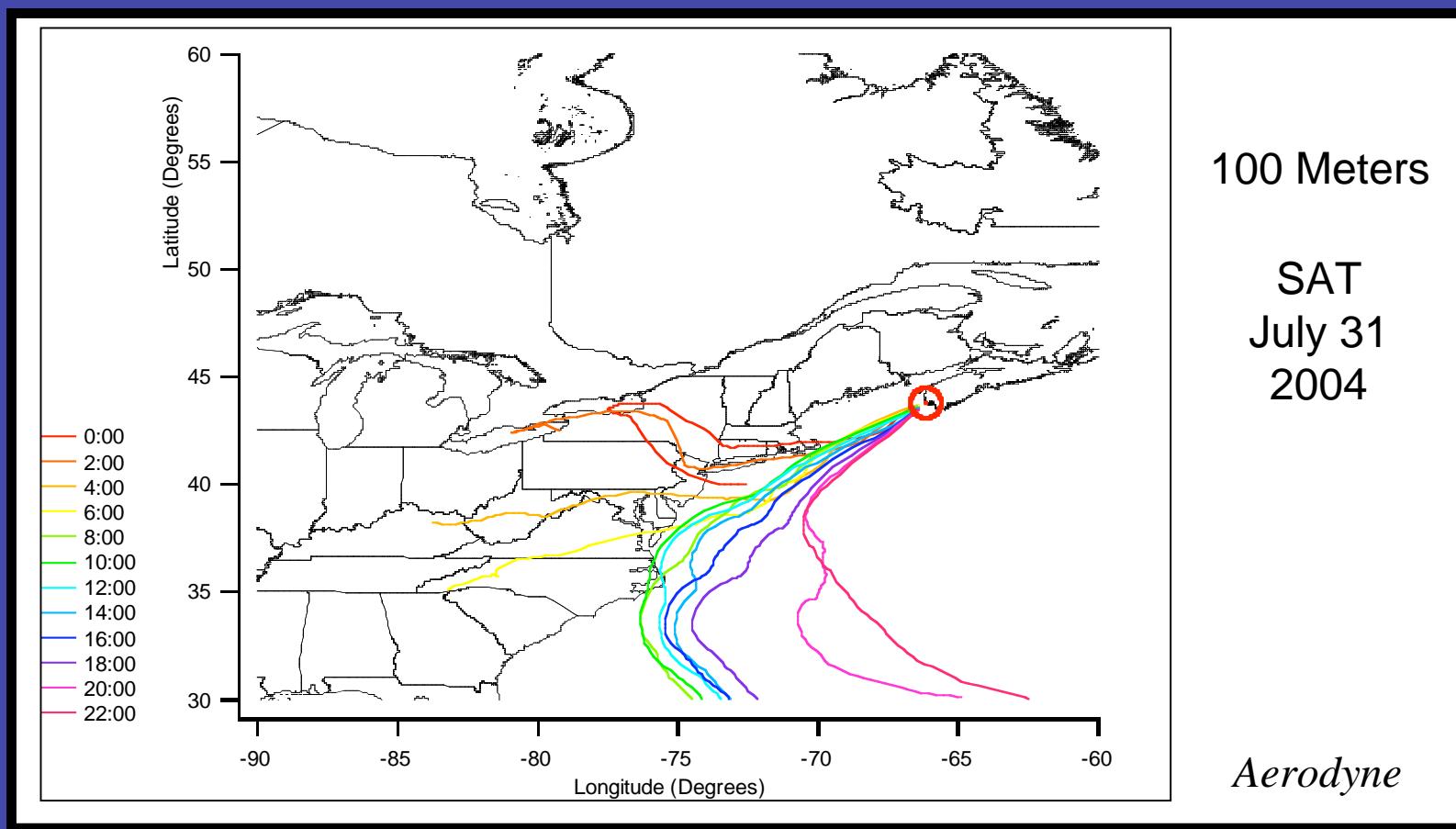
Eastern U.S. Emissions

F1 (U.S.) can be further split:

- F1a: 1° & 2° gas-phase & particle phase pollution
- F1b: Very high in particulate SO_4 , NH_4 , OC, and O_3

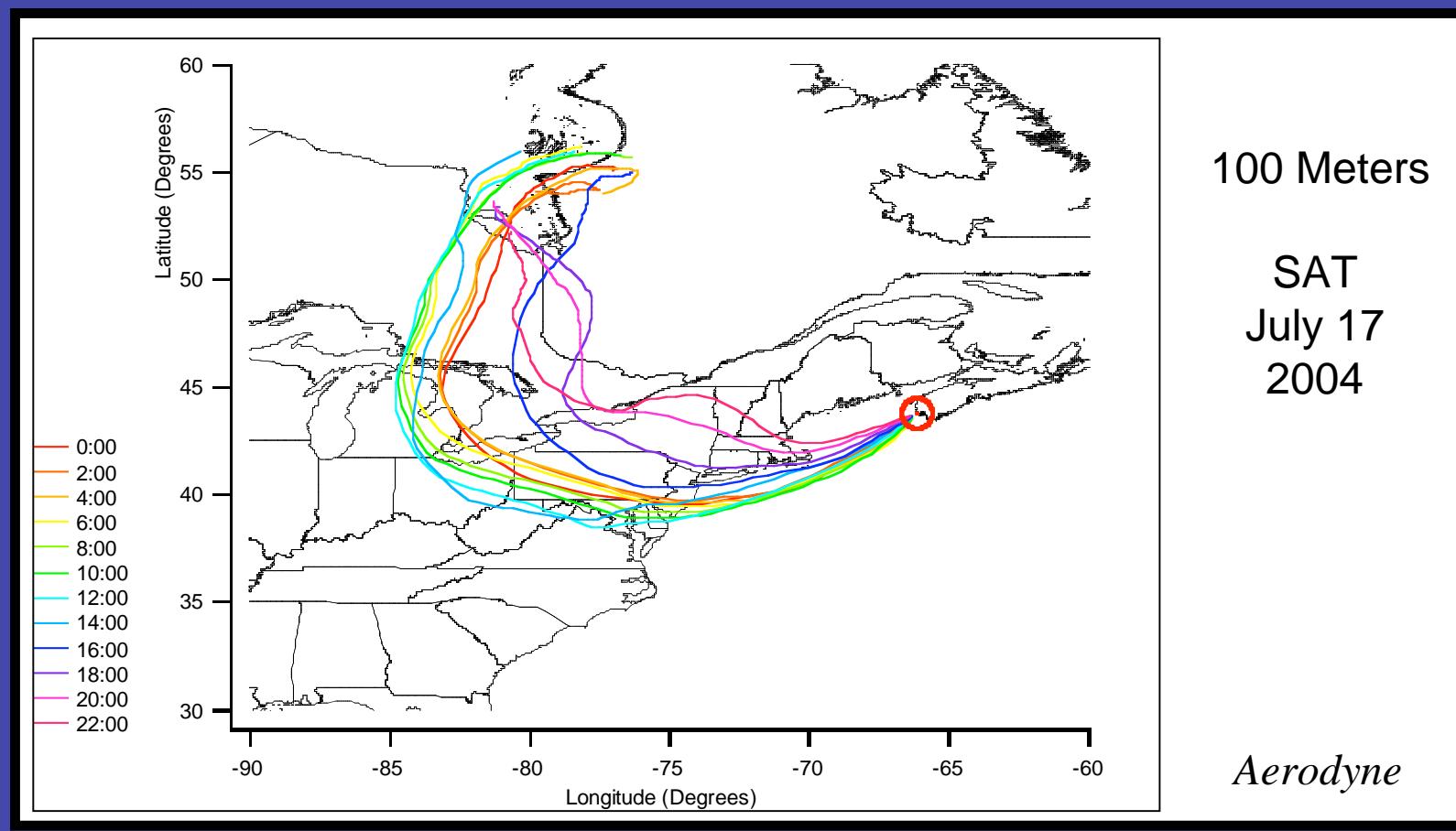
Eastern U.S. Emissions

Example 4d back-trajectory for Fla:



Eastern U.S. Emissions

Example 4d back-trajectory for F1b:



Organic Aerosol Chemical Composition

- Total aerosol OC variability driven by:
 - F1a & F1b (US emissions) (most important)
 - Also F4 (OVOCs) & F6 (Biogenic OX)
- Can we get any more detail?

Online Organic Aerosol Chemical Composition

TAG

(B. Williams, A. Goldstein, S. Hering, N. Kreisberg)

- Individual species provide information on OC source types
 - Insights into OC sources and production mechanisms
- All compounds measured appear highly oxygenated
- Identification and quantification: major challenge when far from sources

Organic Aerosol Chemical Composition

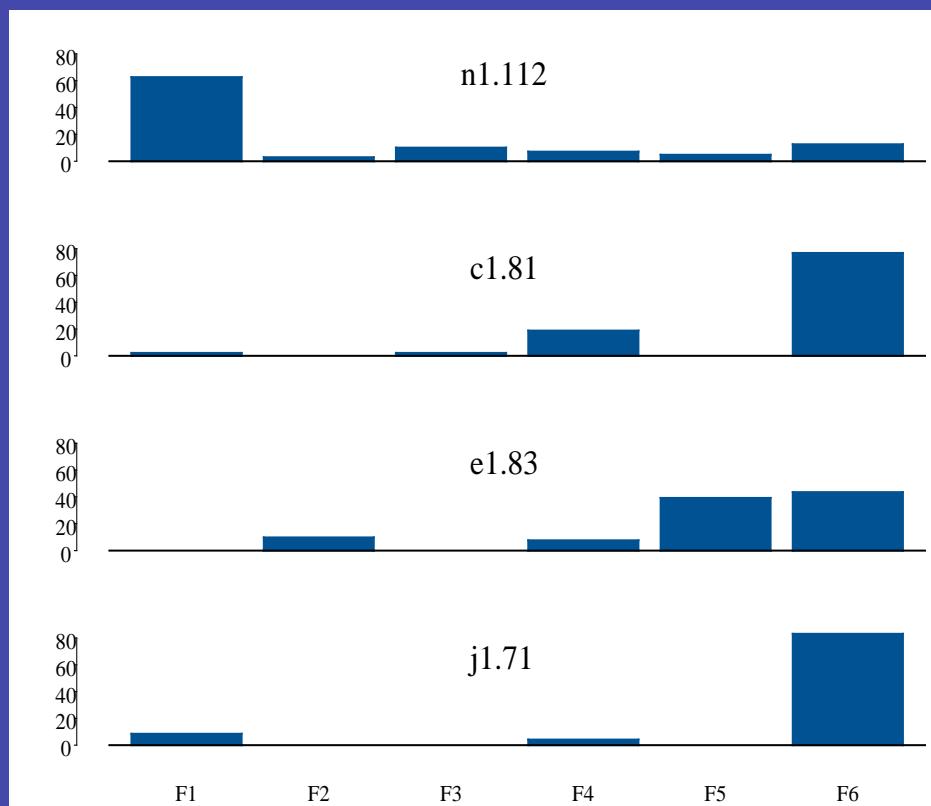
Multiple regression of OC species with 6 factors:

Tentative Compound ID:

1,6-dioxaspiro[4,4]nonane-
2,7-dione ($C_7H_8O_4$) ?

7-anti-methyl-2-oxo-bicyclo[2.2.1]heptane-
7-carboxylic acid ($C_9H_{12}O_3$) ?

2,3-pinane diol ($C_{10}H_{18}O_2$) ?



U.S.

Alkanes

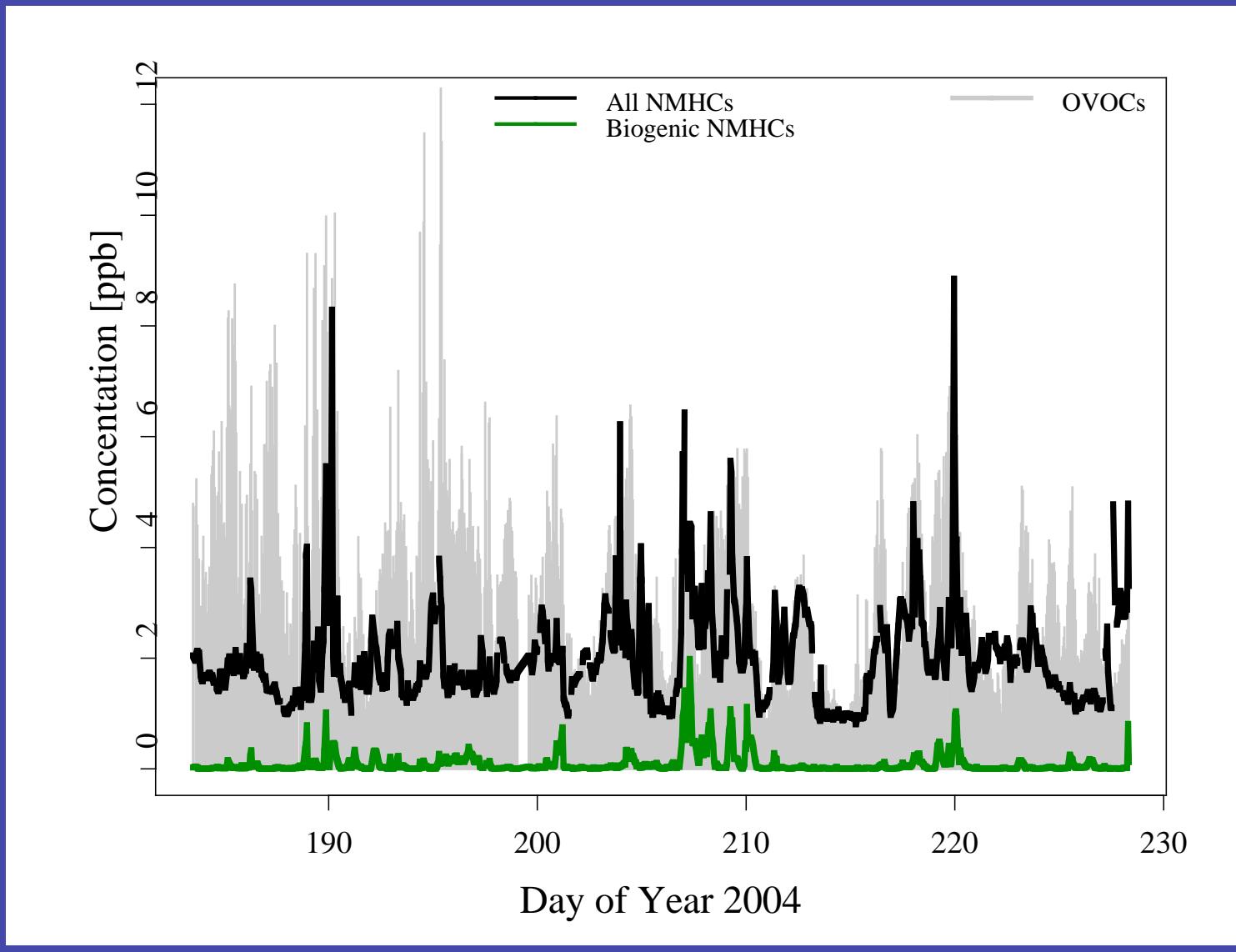
Terpenes

Local comb

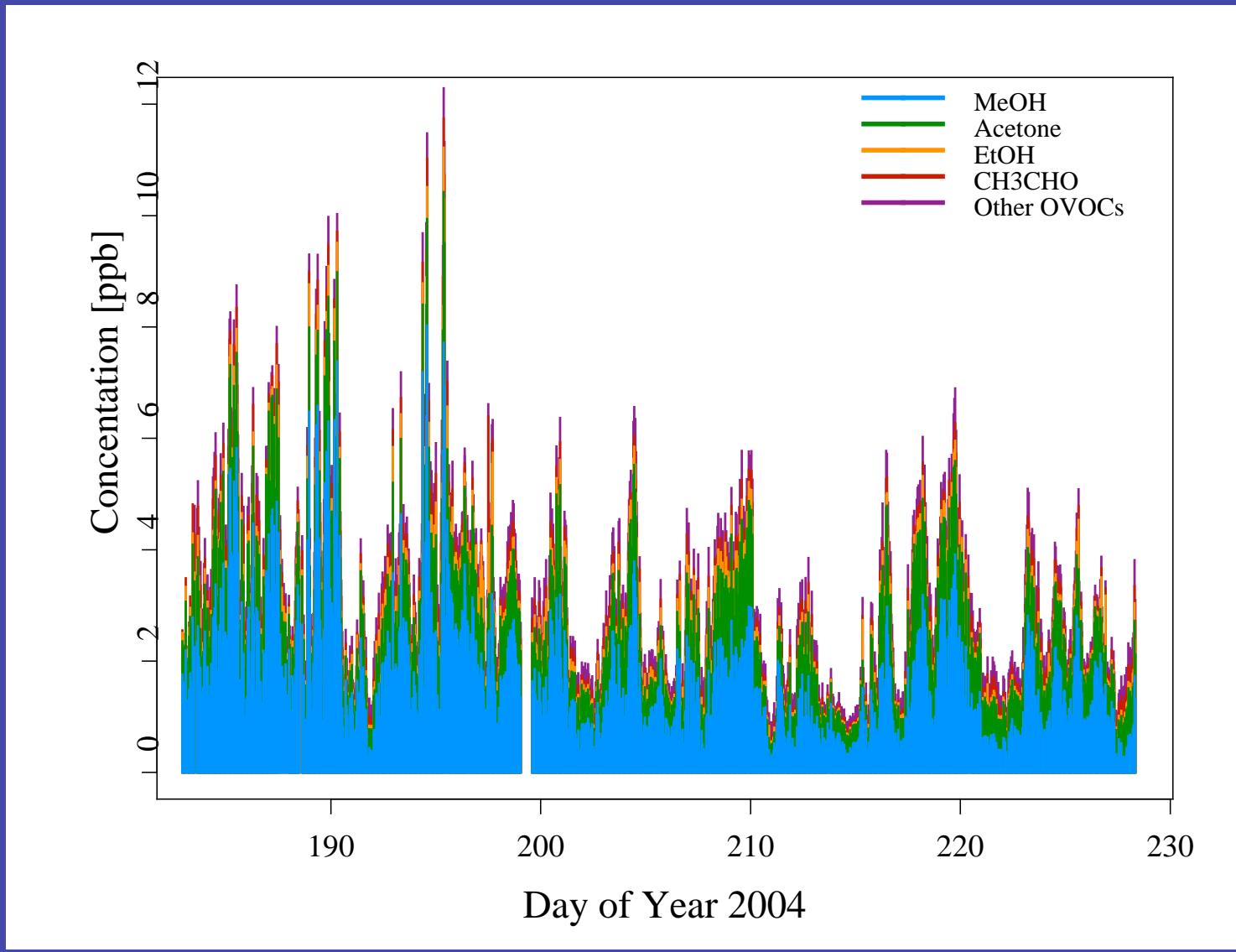
OVOC

Biog OX

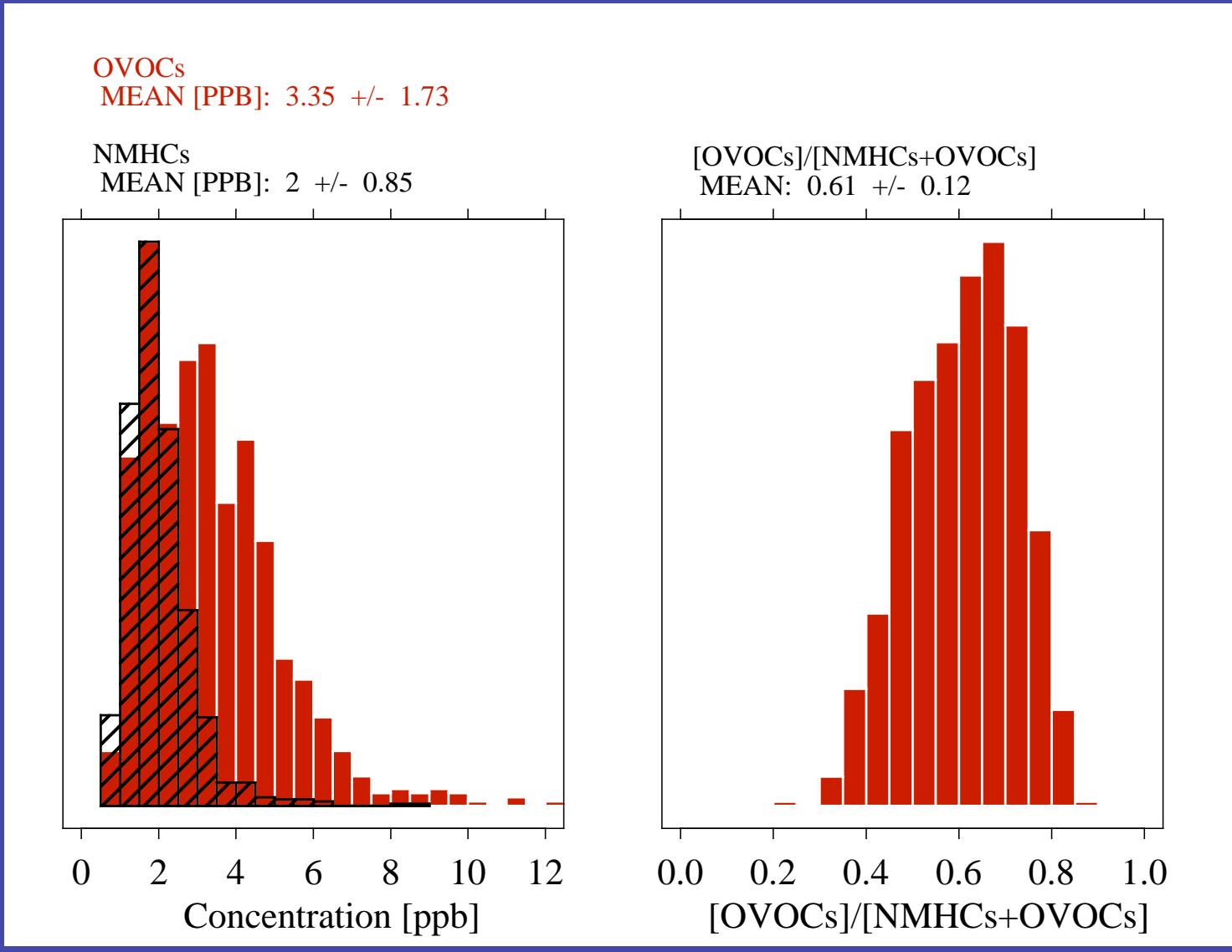
VOC concentrations



What About the OVOCs?



What About the OVOCs?

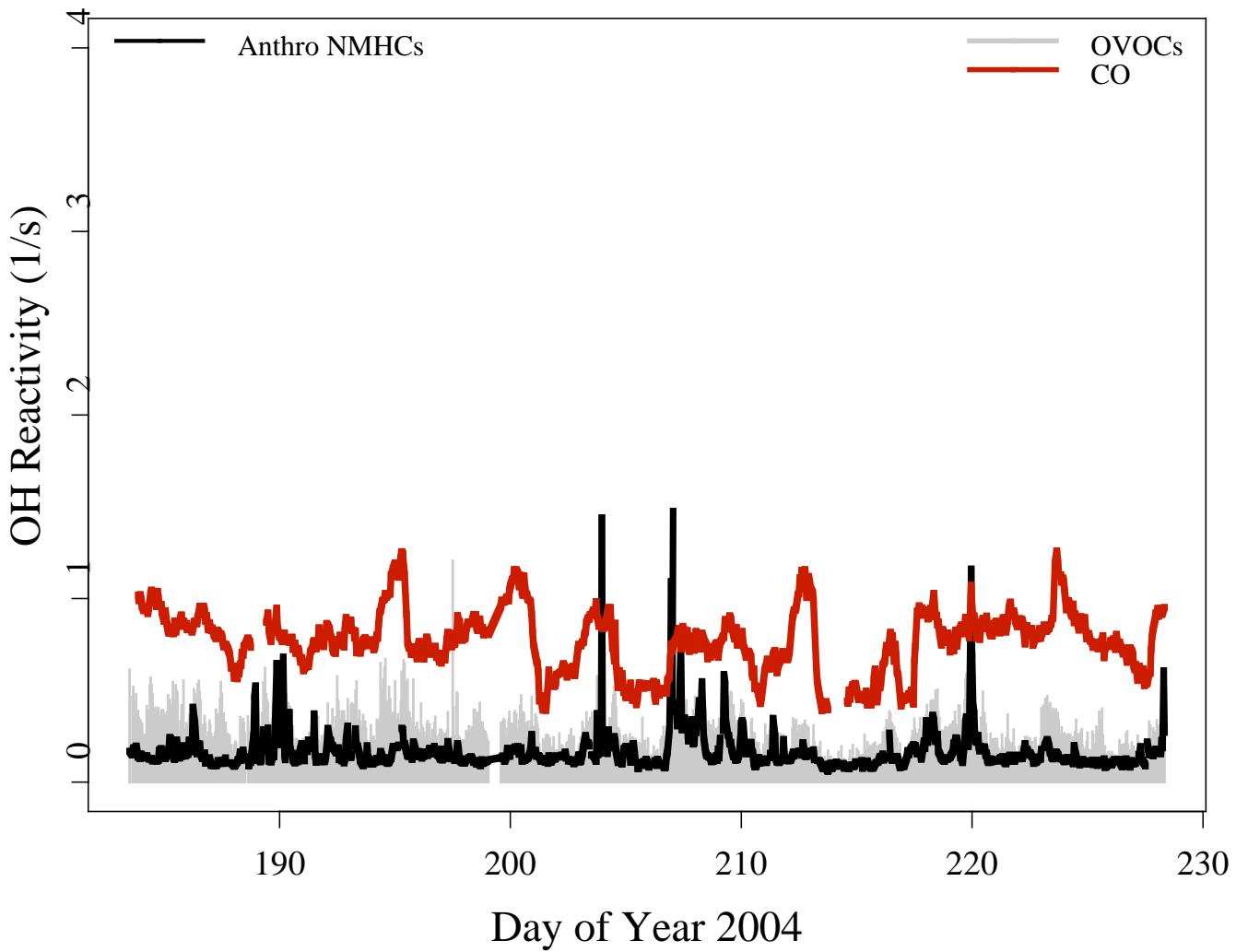


VOC OH Reactivity

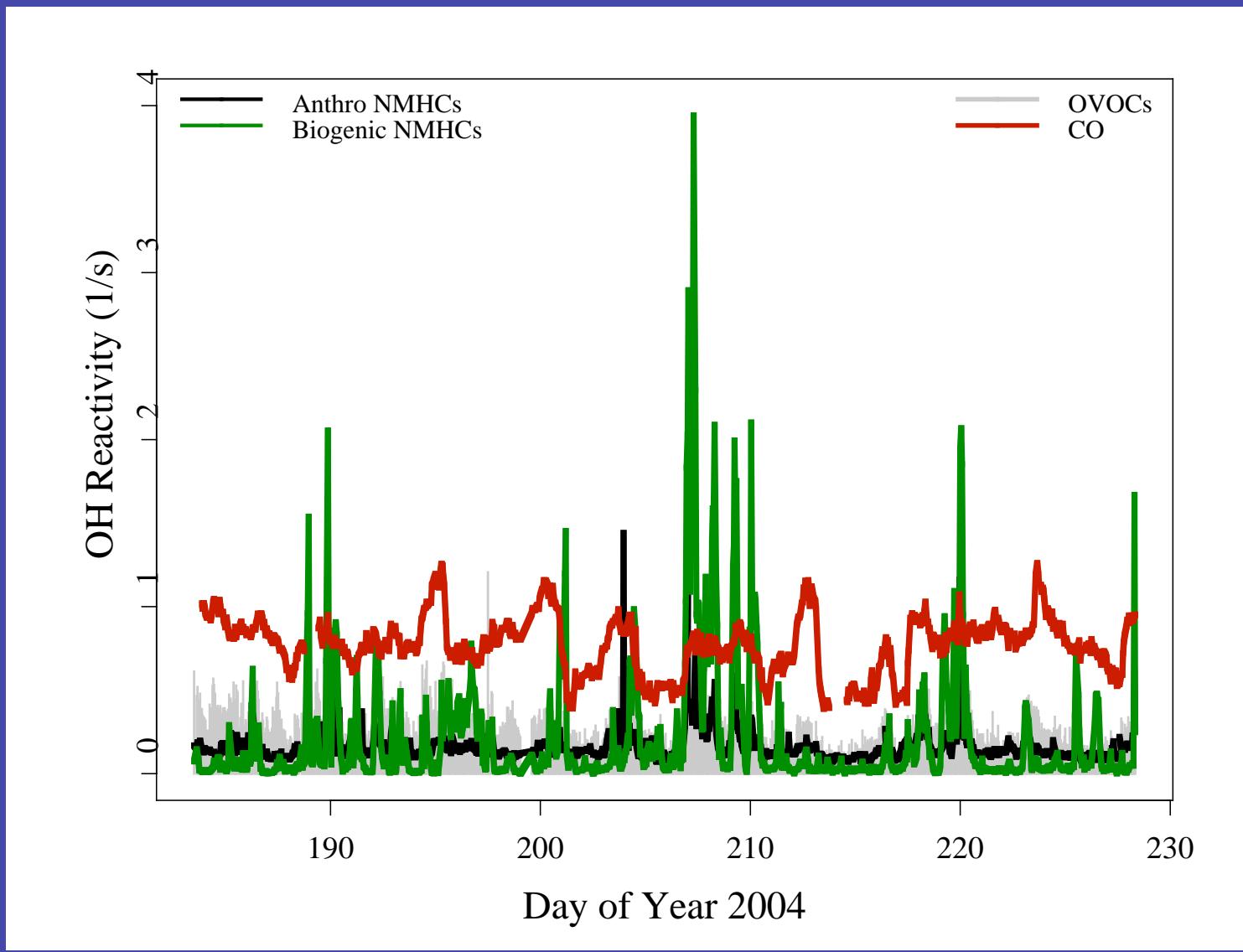
$$L_{OH} = \sum_i k_i [X_i]$$

- Index of the chemical reactivity of an air mass
- Inverse lifetime of OH wrt reaction with the measured compounds
- What are the dominant VOCs competing for OH radicals?

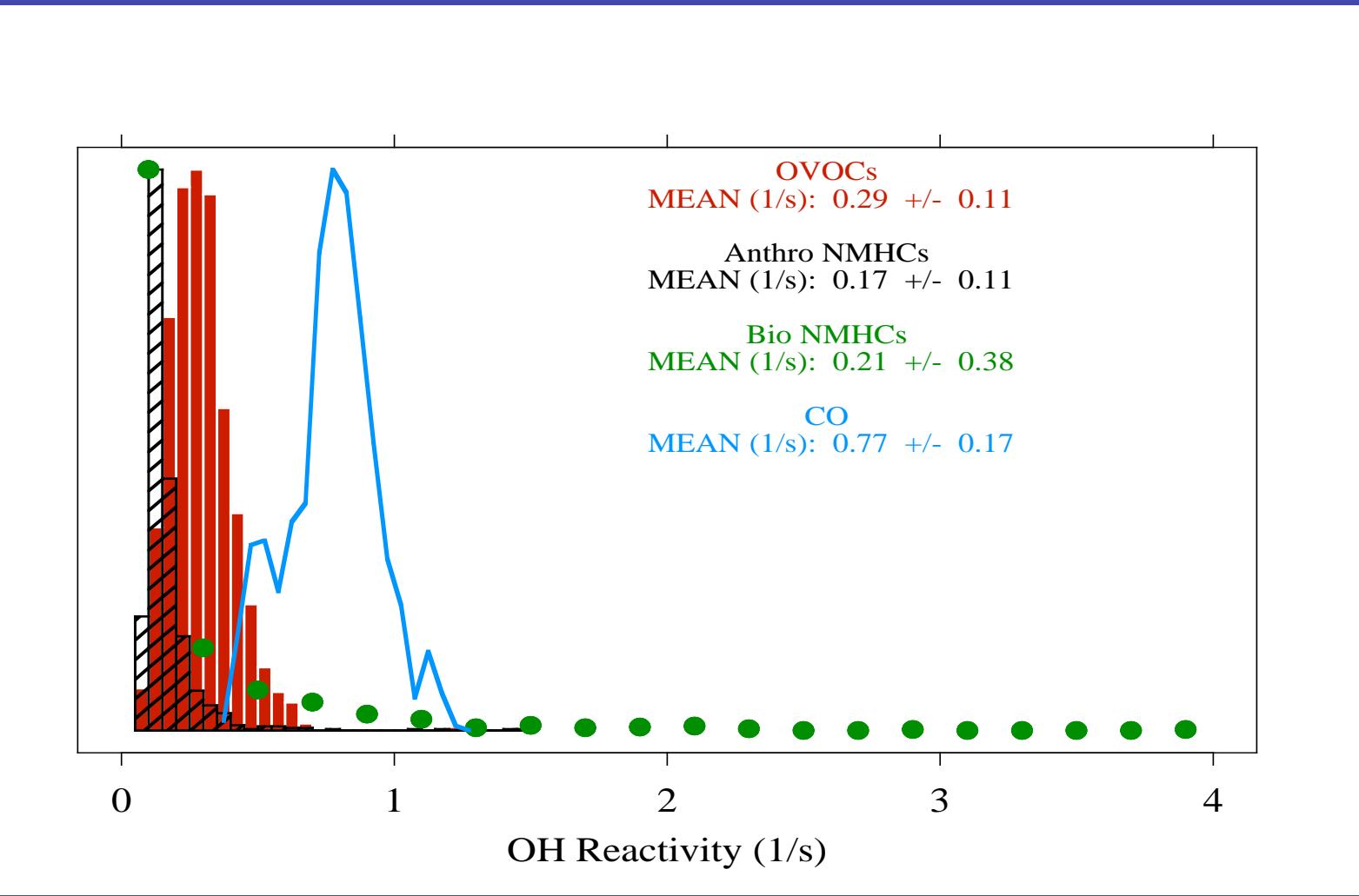
OH Reactivity



OH Reactivity



OH Reactivity

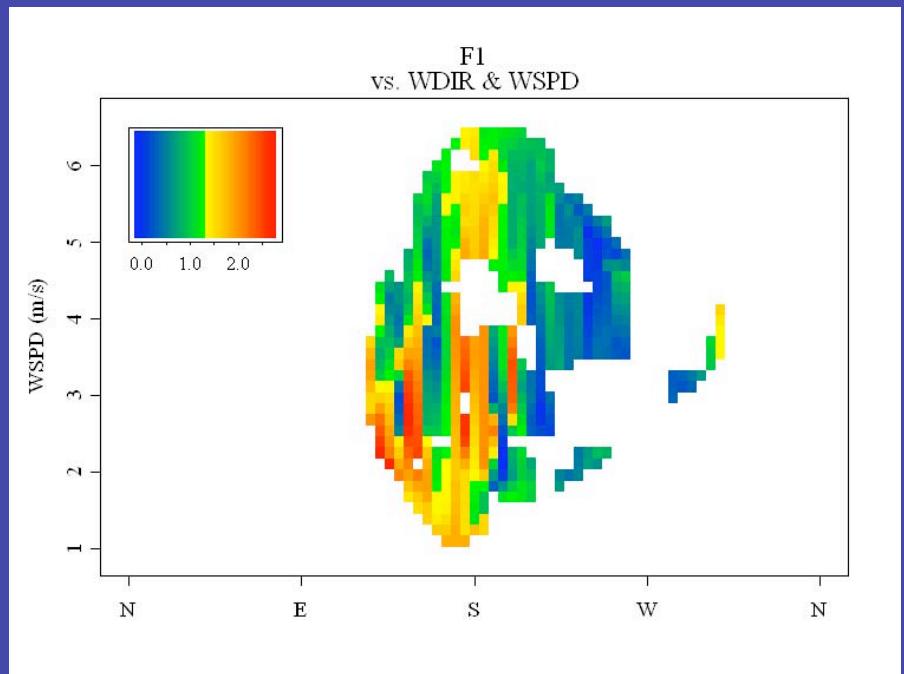


Filtering for NE US emissions

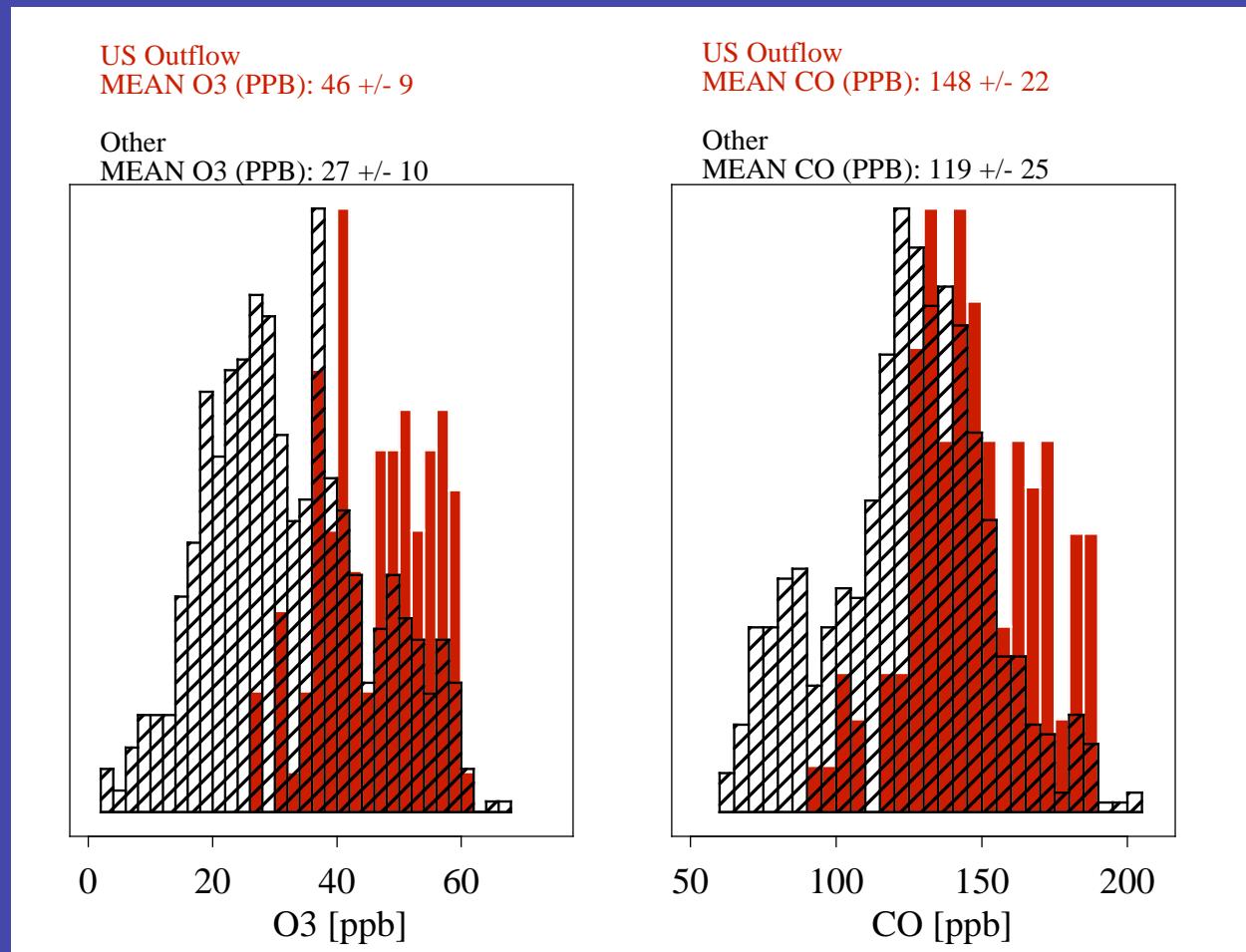
- Use F1 as a filter for NE US outflow
- Choose times when

$$F1 > \sum_{i=2}^6 F_i$$

(117 / 899 hours; 13%)



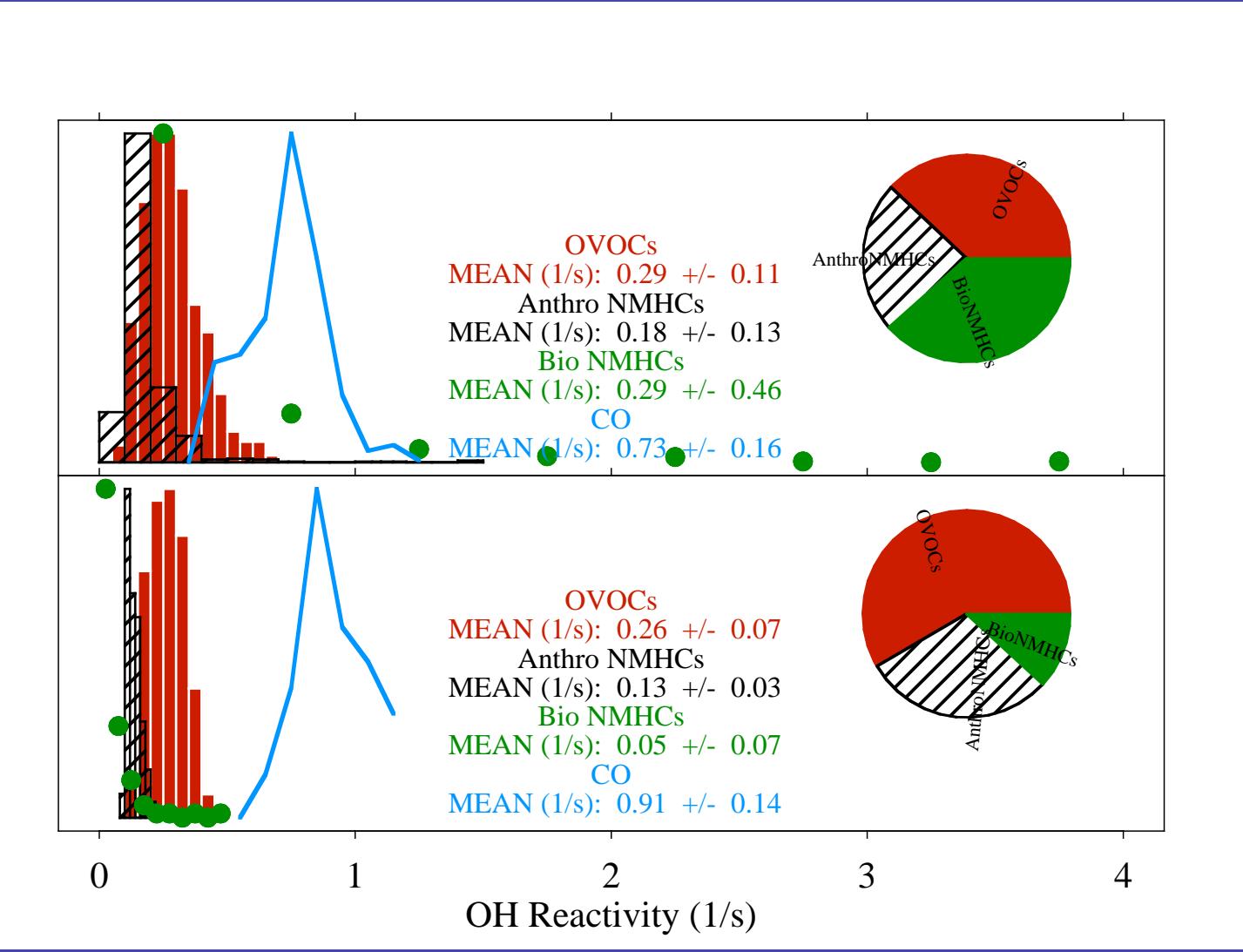
O₃ & CO



Other
periods

US
outflow

VOC OH reactivity



Conclusions & Future Work

Conclusions:

1. CO & O₃ levels similar to 1993
2. FA provides context & means to segregate data
3. Organic aerosol highly oxygenated
 - NE pollution
 - Oxidation of biogenic precursors
4. OVOCs are ~ 2x as important as NMHCs during NE outflow periods

To do:

- Look in detail for evidence of changes in composition & photochemistry since 1993
- Continue investigation of organic carbon budget and chemistry in gas + particle phase

Acknowledgements

Dylan Millet, Brent Williams, Rupert Holzinger, Jen Murphy, Megan McKay (UC Berkeley)

Susanne Hering, Nathan Kreisberg (Aerosol Dynamics)

James Allan, Doug Worsnop, Jose Jimenez (UMIST, Aerodyne, U. Colorado)

Dan Jaffe (UW Bothell)

Funding:

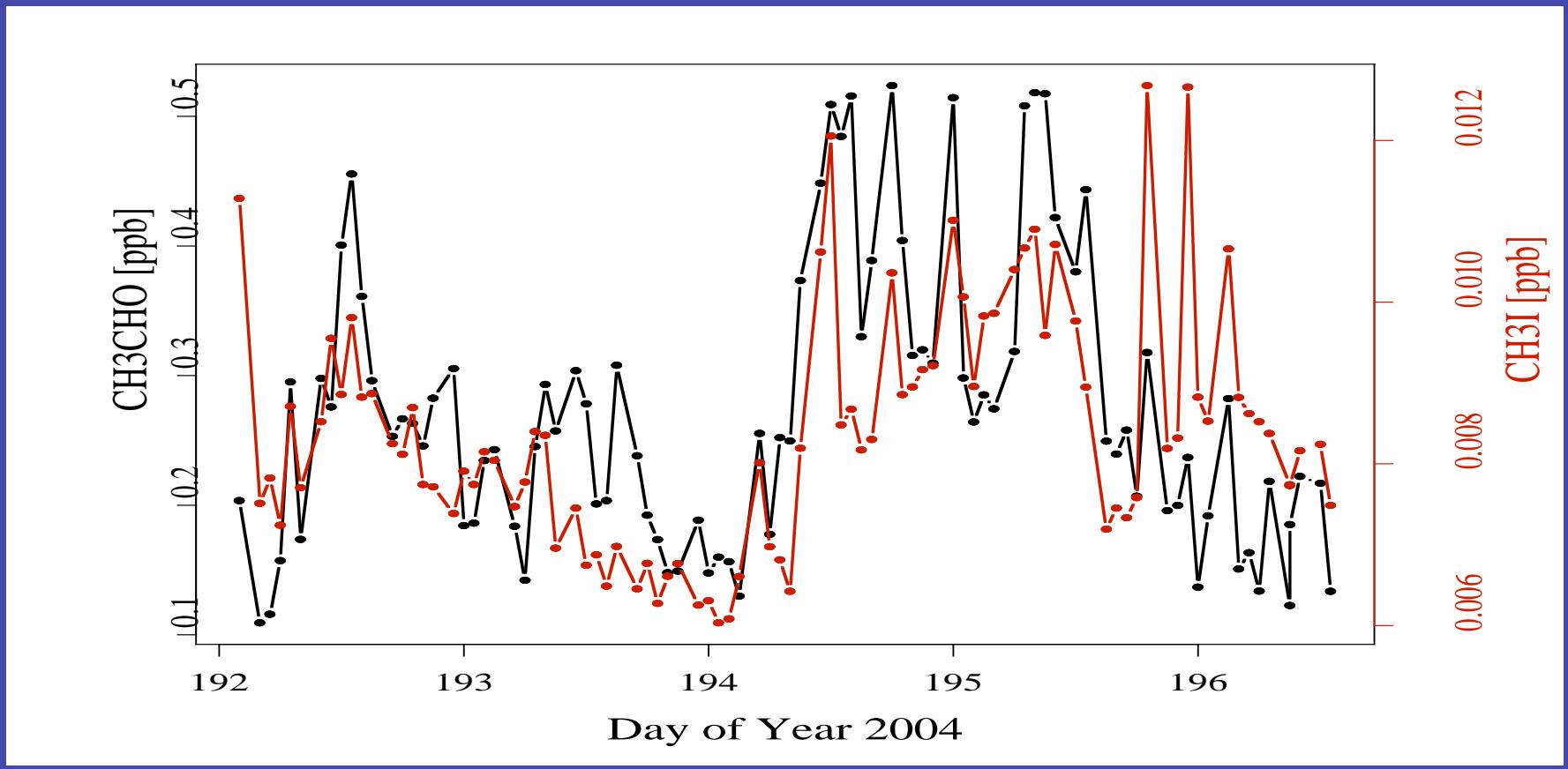
NOAA Office of Global Programs

DOE SBIR Program

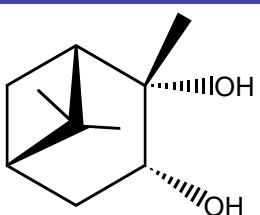
DOE Global Change Education Program

Oceanic OVOC Source?

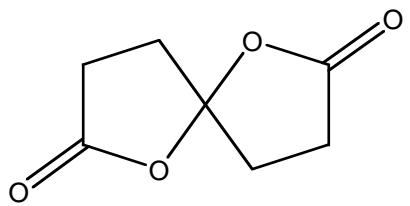
Nothing definitive, but a few hints...



Aerosol OC Composition



j1.71: 2,3-pinane diol



n1.112: 1,6-dioxaspiro[4,4]nonane-2,7-dione